Name: _		
Period: _	Subject:	
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## Nomenclature Sheet #1 (Ionic Compounds)

Give the appropriate chemical formulas for the following compounds. Be sure to include the proper subscripts for each atom.

L		
	1. Iron(III) sulfide	
	2. Copper(I) oxide	
	3. Magnesium chloride	
	4. Potassium bromide	
	5. Lithium carbonate	
	<b>6.</b> Beryllium chloride	
	7. Magnesium nitride	
	8. Lead(IV) phosphate	
	9. Ammonium chloride	
Ĺi	10. Calcium fluoride ist the name of each of the compounds	s represented by the following chemical formula
Li		s represented by the following chemical formula
Li	ist the name of each of the compounds 11. $NH_4NO_3$	s represented by the following chemical formula
Li	ist the name of each of the compounds  11. NH <sub>4</sub> NO <sub>3</sub> 12. Ca(OH) <sub>2</sub>	s represented by the following chemical formula
Li	ist the name of each of the compounds 11. $NH_4NO_3$	s represented by the following chemical formula
Li	ist the name of each of the compounds  11. NH <sub>4</sub> NO <sub>3</sub> 12. Ca(OH) <sub>2</sub>	s represented by the following chemical formula
Li	ist the name of each of the compounds  11. NH <sub>4</sub> NO <sub>3</sub> 12. Ca(OH) <sub>2</sub> 13. CaCO <sub>3</sub>	s represented by the following chemical formula
Li	ist the name of each of the compounds  11. NH <sub>4</sub> NO <sub>3</sub> 12. Ca(OH) <sub>2</sub> 13. CaCO <sub>3</sub> 14. CuSO <sub>4</sub>	s represented by the following chemical formula
Li	ist the name of each of the compounds  11. NH <sub>4</sub> NO <sub>3</sub> 12. Ca(OH) <sub>2</sub> 13. CaCO <sub>3</sub> 14. CuSO <sub>4</sub> 15. Fe <sub>2</sub> O <sub>3</sub>	s represented by the following chemical formula
Li	ist the name of each of the compounds  11. NH <sub>4</sub> NO <sub>3</sub> 12. Ca(OH) <sub>2</sub> 13. CaCO <sub>3</sub> 14. CuSO <sub>4</sub> 15. Fe <sub>2</sub> O <sub>3</sub> 16. CrS	s represented by the following chemical formula
Li	ist the name of each of the compounds  11. NH <sub>4</sub> NO <sub>3</sub> 12. Ca(OH) <sub>2</sub> 13. CaCO <sub>3</sub> 14. CuSO <sub>4</sub> 15. Fe <sub>2</sub> O <sub>3</sub> 16. CrS  17. BeBr <sub>2</sub>	s represented by the following chemical formula
Li	ist the name of each of the compounds  11. NH <sub>4</sub> NO <sub>3</sub> 12. Ca(OH) <sub>2</sub> 13. CaCO <sub>3</sub> 14. CuSO <sub>4</sub> 15. Fe <sub>2</sub> O <sub>3</sub> 16. CrS  17. BeBr <sub>2</sub> 18. NaF	s represented by the following chemical formula

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## Nomenclature Sheet #2 (Ionic Compounds)

Give the appropriate chemical formulas for the following compounds. Be sure to include the proper subscripts for each atom.

1. Iron(IV) nitride	
2. Copper(II) chloride	
3. Calcium oxide	
4. Magnesium bromide	
5. Sodium nitrate	
<b>6.</b> Potassium chloride	
7. Magnesium phosphide	
8. Iron(III) sulfate	
0 4 : :1	
<b>9.</b> Ammonium oxide	
10. Sodium oxide	nds represented by the following chemical fo
10. Sodium oxide	nds represented by the following chemical fo
<ul><li>10. Sodium oxide</li><li>List the name of each of the compour</li><li>11. NH<sub>4</sub>Cl</li></ul>	nds represented by the following chemical fo
10. Sodium oxide  List the name of each of the compour	nds represented by the following chemical fo
<ul><li>10. Sodium oxide</li><li>List the name of each of the compour</li><li>11. NH<sub>4</sub>Cl</li></ul>	nds represented by the following chemical fo
<ul><li>10. Sodium oxide</li><li>List the name of each of the compour</li><li>11. NH<sub>4</sub>Cl</li><li>12. KOH</li></ul>	nds represented by the following chemical fo
<ul> <li>10. Sodium oxide</li> <li>List the name of each of the compour</li> <li>11. NH<sub>4</sub>Cl</li> <li>12. KOH</li> <li>13. Mg(NO<sub>3</sub>)<sub>2</sub></li> </ul>	nds represented by the following chemical fo
<ul> <li>10. Sodium oxide</li> <li>List the name of each of the compour</li> <li>11. NH<sub>4</sub>Cl</li> <li>12. KOH</li> <li>13. Mg(NO<sub>3</sub>)<sub>2</sub></li> <li>14. Cu(NO<sub>3</sub>)<sub>2</sub></li> </ul>	nds represented by the following chemical fo
<ul> <li>10. Sodium oxide</li> <li>List the name of each of the compour</li> <li>11. NH<sub>4</sub>Cl</li> <li>12. KOH</li> <li>13. Mg(NO<sub>3</sub>)<sub>2</sub></li> <li>14. Cu(NO<sub>3</sub>)<sub>2</sub></li> <li>15. Mo<sub>2</sub>O<sub>3</sub></li> </ul>	nds represented by the following chemical fo
10. Sodium oxide  List the name of each of the compour  11. NH <sub>4</sub> Cl  12. KOH  13. Mg(NO <sub>3</sub> ) <sub>2</sub> 14. Cu(NO <sub>3</sub> ) <sub>2</sub> 15. Mo <sub>2</sub> O <sub>3</sub> 16. NiS	nds represented by the following chemical fo
<ul> <li>10. Sodium oxide</li> <li>List the name of each of the compour</li> <li>11. NH<sub>4</sub>Cl</li> <li>12. KOH</li> <li>13. Mg(NO<sub>3</sub>)<sub>2</sub></li> <li>14. Cu(NO<sub>3</sub>)<sub>2</sub></li> <li>15. Mo<sub>2</sub>O<sub>3</sub></li> <li>16. NiS</li> <li>17. KBr</li> </ul>	nds represented by the following chemical fo
<ol> <li>Sodium oxide</li> <li>List the name of each of the compour</li> <li>NH<sub>4</sub>Cl</li> <li>KOH</li> <li>Mg(NO<sub>3</sub>)<sub>2</sub></li> <li>Cu(NO<sub>3</sub>)<sub>2</sub></li> <li>Mo<sub>2</sub>O<sub>3</sub></li> <li>NiS</li> <li>KBr</li> <li>BaI<sub>2</sub></li> </ol>	nds represented by the following chemical fo